## CHE 102 - Homework - Ch 20a Naming and Drawing Alkenes and Alkynes

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Name:	Date:	
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[4 pt] 1. Complete the following table:

Property	Alkenes	Alkynes	
Functional Group			
Naming "ending"			
Location (Y/N)			
Location (1/1)			
Shape/Bond Angle			

[6 pt] 2. Define the term Geometric Isomer (cis-trans isomer). What are the **TWO** requirements for a molecule to be capable of having cis-trans isomers? Draw **AND** label two molecules illustrating a cis and a trans isomer (**DO NOT** use the examples given in class or the book, come up with your own example).

[5 pt] 3. Which of the following molecules have structural formulas that permit cis-trans isomers to exist (Yes/No). Explain your answers.

(a) 
$$CCl_2=CBr_2$$

(c) 
$$CH_2 = CHCl$$

(d) 
$$CH_3CH_2C \equiv CCH_3$$

(e) 
$$(CH_3)_2C = CHCH_3$$

[10 pt] 4. Name the following molecules using IUPAC nomenclature:

(a) CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CH=CHCH <sub>3</sub>	(b) Br Br
(c)	(d)
(e)	(f) CI
(g)	(h)
(i)	(j)

## CHE 102 - Homework - Ch 20a

[10 pt] 5. Draw the following molecules using Lewis Structure or Line Drawings:

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(a) 2-chloro-3-octene	(b) 4-methyl-1-heptyne
(c) 4-ethyl-2-hexyne	(d) 6-methyl-3-heptene
(e) 3-ethyl-4-propyl-2-nonene	(f) 3-bromo-4-methyl-2-hexene
(g) 3-ethyl-5,5-dimethyl-2-hexene	(h) 2-bromopentane
(i) cis-2-pentene	(j) trans-2-pentene

[10 pt] 6. Name the following molecules using IUPAC nomenclature:

(a) CI Br	(b) CI F
(c) OH	(d) NH <sub>2</sub>
(e) O	(f) Br
(g) O CI	(h)
(i)	(j) CI

## CHE 102 - Homework - Ch 20a

[10 pt] 7. Draw the following molecules using Lewis Structure or Line Drawings:

(a) p-chlorobenzaldehyde	(b) nitrobenzene
(c) p-ethylphenol	(d) o-bromoaniline
(e) 1,5-diphenyl-3-heptyne	(f) p-ethylpropylbenzene
(g) 3-ethyl-4-phenyl-2-heptene	(h) m-ethylbenzaldehyde
(i) m-chloroethylbenzene	(j) m-nitrobenzoic acid